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column. (Sample answers: molar mass, empirical formula, percentage composition) Then, have students list characteristics of the compounds that can differ in the second column. (Sample answers: structural formula, mass spectrum, melting point, and boiling point) • Have students complete Investigation 6.9.1. Applicable

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is reversible. Ask, Would the reactants ever be completely used up in a reversible reaction? Why or why not? (No. Some of the product that forms turns back into reactants, so there are always some reactants still present.)

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This site provides examples of current projects underway using green chemistry. Green Chemistry Program Page (244, #4 ) This page explains the EPA's policies and current projects with green chemistry. You can also submit requests for further information on a particular project or topic.

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### CHAPTER 3 CHEMICAL REACTIONS

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### 5.1 QUANTITATIVE ANALYSIS

=11 m/s Statement: The final speed of each person after the collision is 11 m/s. Section 5.3 Questions, page 239 1. Answers may vary. Sample answer: (a) Since the boxes stick together after the collision, we know this is an inelastic collision. Momentum is conserved in an inelastic collision. Momentum is always conserved if there are no

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