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Section 15-5 Spontaneous Processes A spontaneous process is a physical or chemical change that once begun, occurs with no outside intervention. Many spontaneous processes require some energy from the surroundings to start the process. 53.

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See an explanation and solution for Chapter 15, Problem 15.21 in Amateis/Silberberg's Chemistry: The Molecular Nature of Matter and Change (8th Edition).

[Solved] Chapter 15, Problem 15.21 - Chemistry: The ...

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Figure 15.4.1 Energy Changes Accompanying the Thermite Reaction Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe₂O₃(s) is -851.1 kJ, whether the reaction occurs in a single step (ΔH 4, shown on the left) or in three hypothetical steps (shown on the right) that involve the successive formation of solid Al₂O₃ and ...

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